

23 Useful Formulae

23.1 Inverse Square Law

D_1 = original distance from source

D_2 = new distance from source

R_1 = original radiation field

R_2 = new radiation field

$$R_2 = R_1 \frac{(D_1)^2}{(D_2)^2}$$

e.g., if a source is 10 mrem/h at 5 feet, then at 13 feet it is

$$10 \times \frac{5^2}{13^2} \text{ mrem/h} = 1.5 \text{ mrem/h.}$$

23.2 Air Sample Calculations

Particulates (complete calculation):

Sample collection time = T minutes

Collection flow rate = F m³/minute (usually 0.7)

Counter efficiency = E%

Background count rate = A₁ CPM

Count rate with sample in tray = A₂ CPM

$$DPM m^3 = \frac{(A_2 - A_1) 100}{E} \times \frac{1}{TF}$$

23.3 Radioactive Decay

The following symbols will be used in this section:

N_0	=	number of unstable nuclei at some original time
N	=	number of unstable nuclei remaining after a time interval t
I_0	=	intensity of radiation at some original time
I	=	intensity of radiation after a time interval t
A_0	=	activity of sample at some original time
A	=	activity remaining after a time interval t
λ	=	decay constant for particular radioactive element
e	=	base of natural logarithms, 2.718
$T_{1/2}$	=	half-life
$n = t/T_{1/2}$	=	number of half-lives

- (1) $N = N_0 e^{-\lambda t}$ or $N = N_0 e^{-0.693t/T_{1/2}}$
 (2) $A = A_0 e^{-\lambda t}$ or $A = A_0 e^{-0.693t/T_{1/2}}$
 (3) $I = I_0 e^{-\lambda t}$ or $I = I_0 e^{-0.693t/T_{1/2}}$
 (4) $N = N_0 2^{-n}$ or $N/N_0 = 1/2^n$

23.4 Decay Constant

$$\lambda = 0.693/T_{1/2}$$

23.5 Specific Activity

$$\text{Specific activity} = \lambda N = 0.693 N/T_{1/2} = \text{dis/sec/g}$$

where $T_{1/2}$ = half-life in seconds

N = number of atoms per gram

$$\text{Specific activity} = \lambda N (3.7 \times 10^{10}) = \frac{N \times 1.873 \times 10^{-11}}{T_{1/2}} = \text{Ci/g}$$

23.6 Radiation Absorption

Alpha Particle Range

For $E < 4$ Mev:

$$R_{\alpha} = 0.56 E$$

For $4 < E < 8$ MeV

$$R_{\alpha} = 1.24 E - 2.62$$

where R_{α} = range in cm of air at 1 atm and 15°C
 E = energy in Mev

Beta Particle Range

For $0.01 \leq E \leq 2.5$ Mev

$$R = 412 E^{1.265 - 0.00541 nE},$$

$$\ln E = 6.63 - 3.2376(10.2146 - \ln R)^{1/2},$$

where R = range in mg/cm².
 E = maximum energy in Mev.

For $E \geq 2.5$ MeV

$$R = 530 E - 106,$$

Where R and E are the same as above.

Sargent's Rule ($E > 0.8$ Mev)

$$R = 0.0526E - 0.094,$$

where R = range in g/cm^2 ,
E = maximum energy in Mev.

Feather's Rule ($E > 0.6$ Mev)

$$R = 0.542 - 0.133,$$

where R and E are the same as for Sargent's Rule.

Gamma Ray Absorption

The following symbols will be used in this section

I_0	=	original radiation exposure rate
I	=	attenuated radiation exposure rate
μ	=	linear absorption coefficient (cm^{-1}) = $\frac{0.693}{x_{1/2}}$
μ/ρ	=	mass absorption coefficient (cm^2/g)
ρ	=	absorber density (g/cm^3)
x	=	absorber thickness (cm)
$x_{1/2}$	=	half-value layer of absorber (cm)
e	=	base of natural logarithms (2.718...)
b	=	"build-up" factor

For monoenergetic or monochromatic narrow-beam radiation,

$$I = I_0 e^{-\mu x}$$

or

$$I = I_0 e^{-(\mu/\rho)(\rho)(x)} ;$$

for monoenergetic or monochromatic wide-beam radiation,

$$I = I_0 e^{-\mu x}$$

Neutron Absorption (for a collimated beam of monoenergetic neutrons)

$$I = I_0 e^{-\sigma N x}$$

- where I_0 = initial neutron intensities,
 I = final neutron intensities,
 N = number of atoms per cc in the absorber,
 σ = cross section (cm^2)
 x = thickness of absorber (cm),
 e = base of the natural logarithm (2.718...).

Since this equation is only an approximation of neutron attenuation, average neutron energies can be used for determining the value of σ . The equation is not accurate enough to justify the use of neutron build up factors.

Approximate Range-Energy Relation for Protons

$$R = (E/9.3)^{1.8}$$

- where E = energy in Mev (few Mev to 200 Mev),
 R = range in meters in air.

23.7 Beta Particle Counting

Self-Absorption

$$\frac{R_0}{R} = \frac{1}{mx} (1 - e^{-mx})$$

$$\frac{R_0}{R} = \frac{1}{mx} (1 - e^{-mx}),$$

- where R_0 = measured counting rate,
 R = true counting rate
 x = sample thickness (mg/cm²)
 m = absorption coefficient (cm²/mg)

Resolving-Time Determination

$$\tau = \frac{R_1 + R_2 - R_{12}}{2(R_1 R_2)}$$

- where τ = resolving time in seconds,
 R_1 = counting rate, source 1 (c/s)
 R_2 = counting rate, source 2 (c/s)
 R_{12} = counting rate, combined sources 1 and 2 (c/s)

Resolving-Time Correction

$$R = \frac{R_0}{1 - R_0 \tau}$$

- where R = true counting rate (c/s),
 R_0 = observed counting rate (c/s),
 τ = resolving time in seconds.

23.8 Calibration Procedure

The equations in this section are applicable to dose in air from gamma emitters.

Exposure Rate from a Point Source

$$I_{\gamma} = 0.156 n E (10^5 \mu_a) ,$$

- where I_{γ} = mr/hr at 1 meter per mCi,
 n = gamma quanta per disintegration,
 E = energy of gamma quanta in Mev,
 μ_a = energy absorption coefficient for gamma in air (STP) in cm^{-1} .

The equation assumes that one ion pair in air causes an average energy expenditure of 32.7 eV.

Approximate Exposure Rate from Any Gamma Point Source

$$r/hr \text{ at 1 foot} = 6CEn ,$$

or

$$mr/hr/mCi \text{ at 1 meter} = 0.5 nE ,$$

- where C = number of curies,
 E = gamma-ray energy in Mev,
 n = gamma quanta per disintegration.

Exposure Rate from Any Gamma Point Source

$$mr/hr = nI_{\gamma} / s^2 ,$$

- where n = number of millicuries,
 I_{γ} = mr/hr at 1 meter per mCi,
 s = distance in meters.

23.9 Internal Radiation Dosage

Biological Half-Life

$$T_b = \frac{0.693}{\lambda_b} ,$$

where λ_b = biological rate of elimination constant,
 T_b = biological half-life

Effective Half-Life

$$T_{eff} = \frac{(T_{1/2})(T_b)}{T_{1/2} + T_b} ,$$

where T_{eff} = effective half life,
 $T_{1/2}$ = radioactive (physical) half-life
 T_b = biological half life.

Beta-Emitter Dose

$$D = 88ET_{eff}C(1 - e^{-\lambda_{eff}t}) ,$$

where D = dose (reps),
 E = average energy of beta particle (Mev)
 C = Ci/g of radioisotope in tissue
 λ_{eff} = effective decay constant (days⁻¹)
 t = time (days).

23.10 Decontamination Factor

$$D.F. = \frac{\textit{Initial Activity}}{\textit{Final Activity}}$$