

C-14 studies at Indian Point

and application in Reg Guide 1.21 reports (1982-2010)



1974

Several papers were presented in the early 1970's identifying C-14 as a legitimate isotope of interest for background studies, as well as emissions from Nuclear Power Plants.

- Bonka, et al, Germany
- Magno, et al, AEC Air Cleaning Conf, San Fran
- Kunz, et al, Health Physics Society, Knoxville
- Dozens of others, many internationally

Production of C-14



PWR assessments prior to 1980

From various works, established production rates were widely accepted:

Reactor Coolant (assuming 7 ppm N ₂)	7 Ci/GW (e) yr
Fuel	15 Ci/GW (e) yr
Structural Material	25 Ci/GW (e) yr

But these did not provide effluent estimates and little or no measurements were made in the US.

1975-1978

- General understandings:
 - Most effluent will be via the main plant vent
 - Most of the dose producing form = CO_2
- Rad Science Institute of NY started sampling and publishing findings (Ginna, Rochester, NY)
 - 11.6 Ci / GW (e) yr
- Many published reports from light water reactors in Germany, indicating an average of 6 Ci/GW(e) yr.

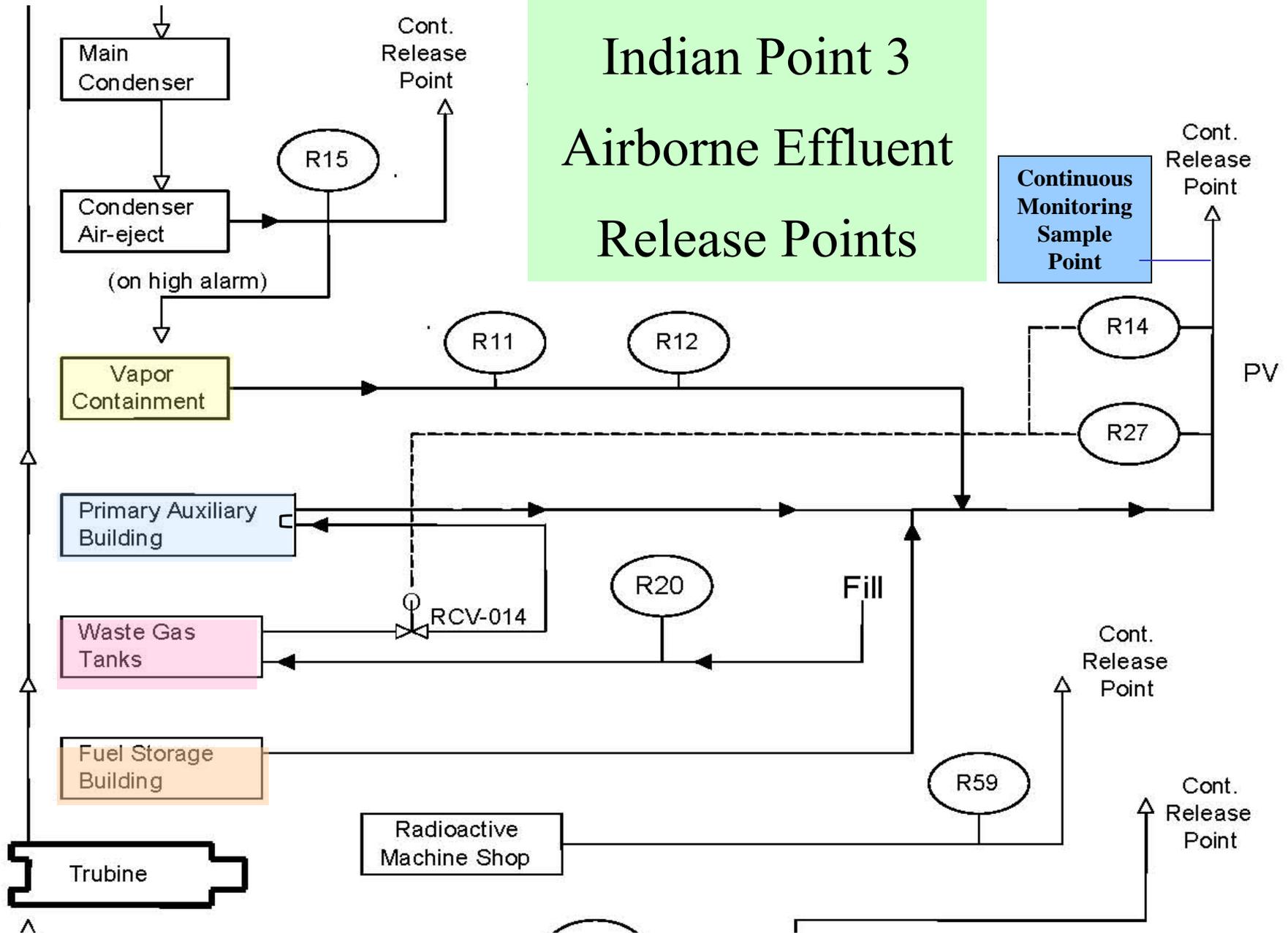
1980, NY Dept of Health

- Charles Kunz was contracted by the NY DOH to sample and report C-14 in effluents from plants owned by the Power Authority State of NY.
 - JA Fitzpatrick, Oswego NY (BWR)
 - Indian Point #3, Westchester Co, NY (PWR)
- Goal was to determine the following:
 - Quantity of C-14 releases,
 - Pathways of release, and,
 - Chemical form of the release.

Sampling at IP #3 (PWR)

- Effluent samplers were operated continuously for 98 weeks, during all phases of operation (including trips and outages).
- Grab samples were collected:
 - Reactor Coolant,
 - Waste water
- Analyzed for inorganic and organic C-14.

Indian Point 3 Airborne Effluent Release Points



Components for Sampling

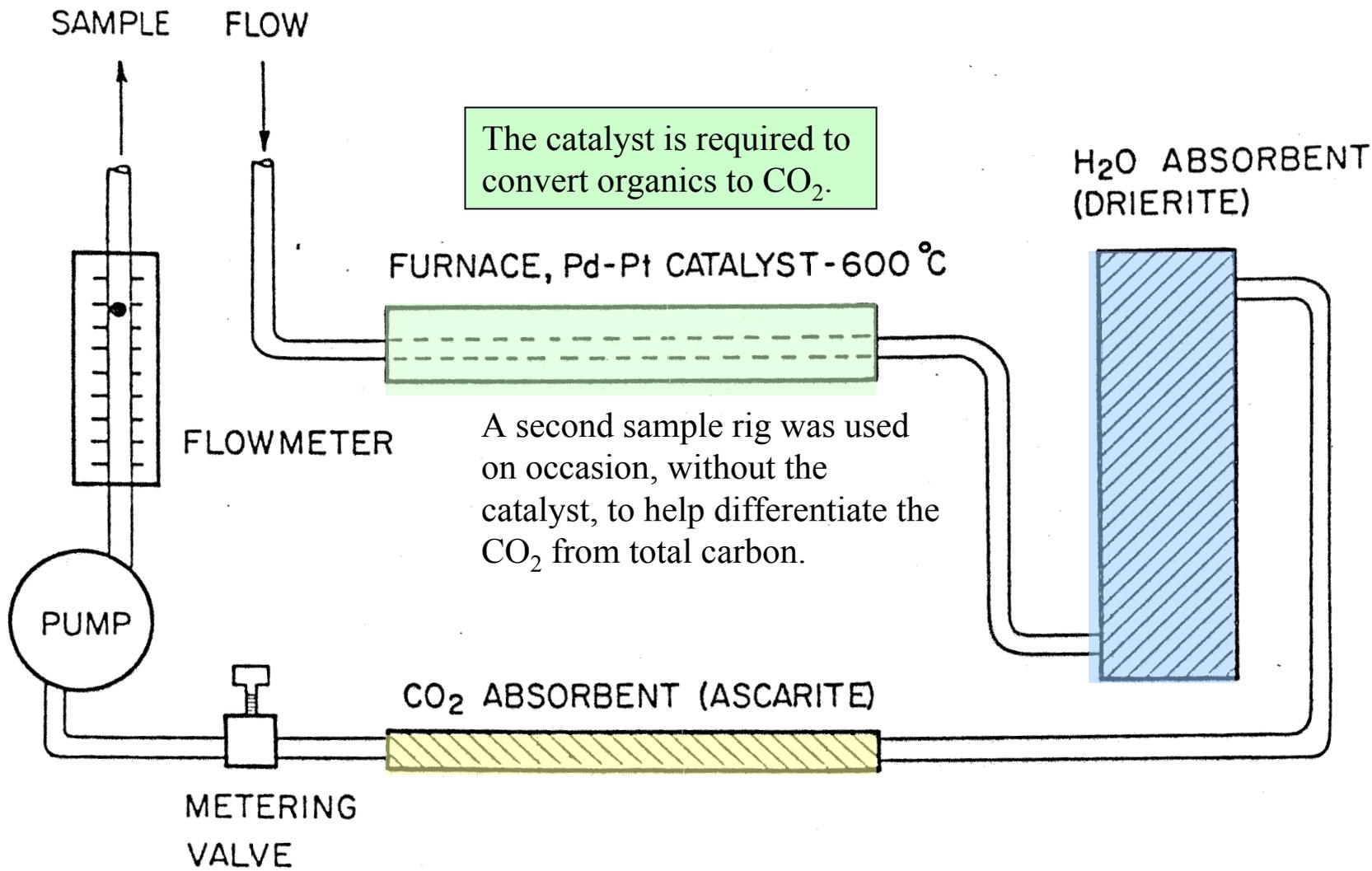
1. Leco Purifying Heater #507-400 (12" long and 2" diameter) and Heating Receptacle #507-015.
2. Furnace Catalyst - 1:1 mixture of Platinum on Alumina and Palladium on Alumina (3.2 mm pellets).
3. Gas drying jar with absorbent (Drierite - indicating, 8 mesh).
4. CO₂ absorbent (Ascarite, 8-20 mesh).
5. Metering valve - Model 22 RS-4, Whitey.
6. Diaphragm pump - Model MB-21, Metal Bellows Company.
7. Flow meter - ball type - Model 7630, Matheson Gas.

Ascarite

Ascarite is a trade name for a form of sodium hydroxide (NaOH) which is coated on non-fibrous silicate (vermiculite). It rapidly and quantitatively absorbs CO₂.

During use, the light brown granules gradually turn white due to absorption of carbon dioxide and formation of sodium carbonate. It is hygroscopic and deliquescent when exposed to water.

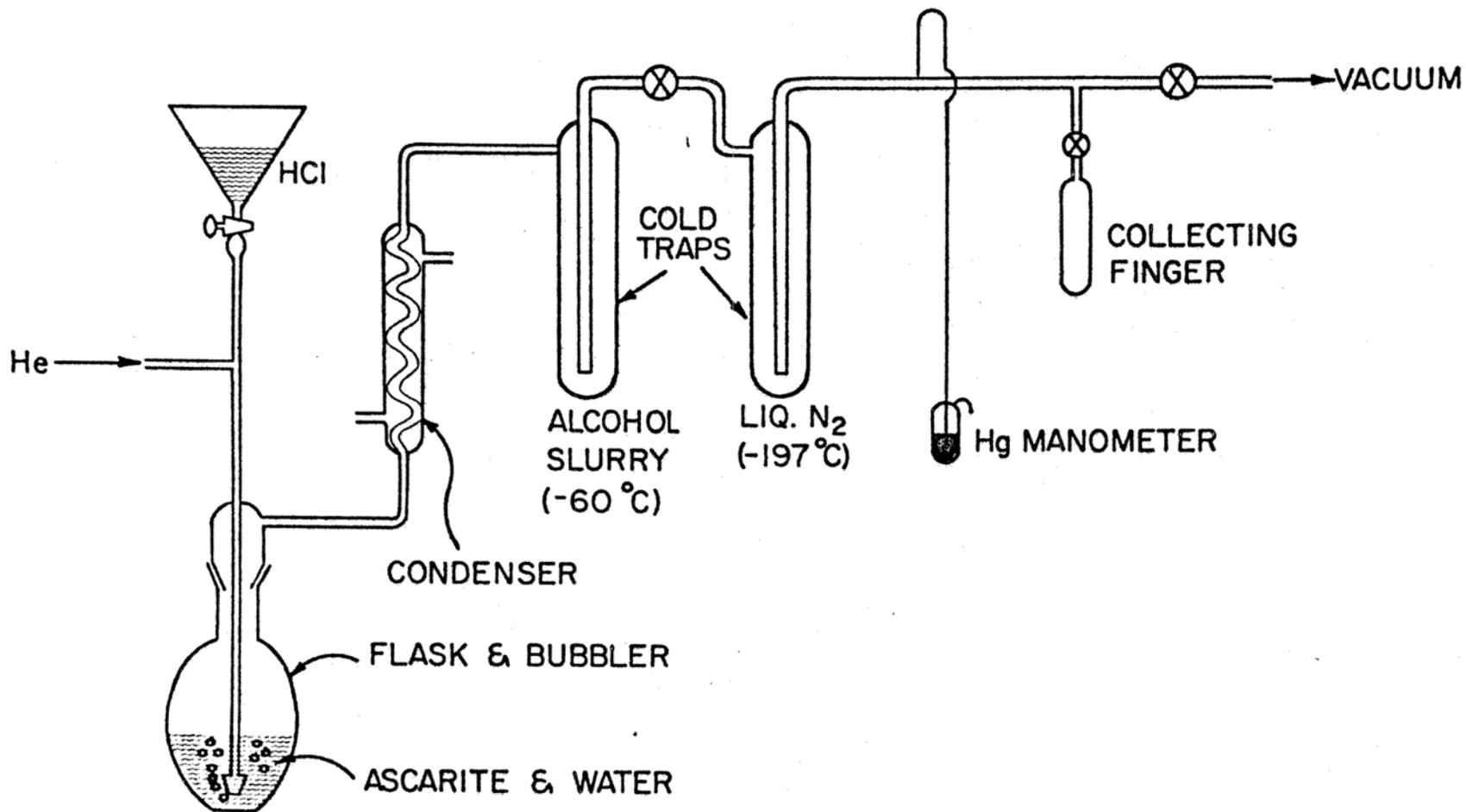
CO₂ absorption capacity is 20 to 30% W/W.



¹⁴C SAMPLER

General sampler basics

- 100 cc/min thru furnace of palladium and platinum alumina.
- Oxygen in the vent system is used to convert methyl, ethyl, CO, etc – for a measure of total carbon, as CO₂.
- Absorbent (silica gel or Drierite) removes water vapor, which could clog the system and/or contaminate it with Tritium.
- Ascarite cartridges (NaOH) were used to absorb (trap) the CO₂ and make Na₂CO₃.
- Columns were changed out weekly and sent to lab.
- Ascarite was acidified, CO₂ extracted, purified, and an aliquot counted on proportional counters.



APPARATUS FOR RECOVERING CO_2 FROM ASCARITE

Summary of Analytical Method

- Ascarite added to 50 mls of water
- Helium was bubbled thru, with 50 mls of concentrated HCl, for approx 30 minutes.
- Gas was condensed thru 2 cold traps to remove water vapor and collect CO₂.
- Volume was measured & an aliquot passed thru a gas chromatograph for purification.
- Loaded into an internal gas proportional tube for counting. (Liquid Scin techniques were not yet fully developed!)

Calculations

- Amount of air thru column determined from flow rate meters or totalizers.
- Sample volume determined by measuring the total CO₂ collected and comparing C-14 concentration to the cc's of CO₂.
- Amount of CO₂ recovered each week was about the same, the capacity of the Ascarite column. Higher when combustion sources were added to the vent gas.

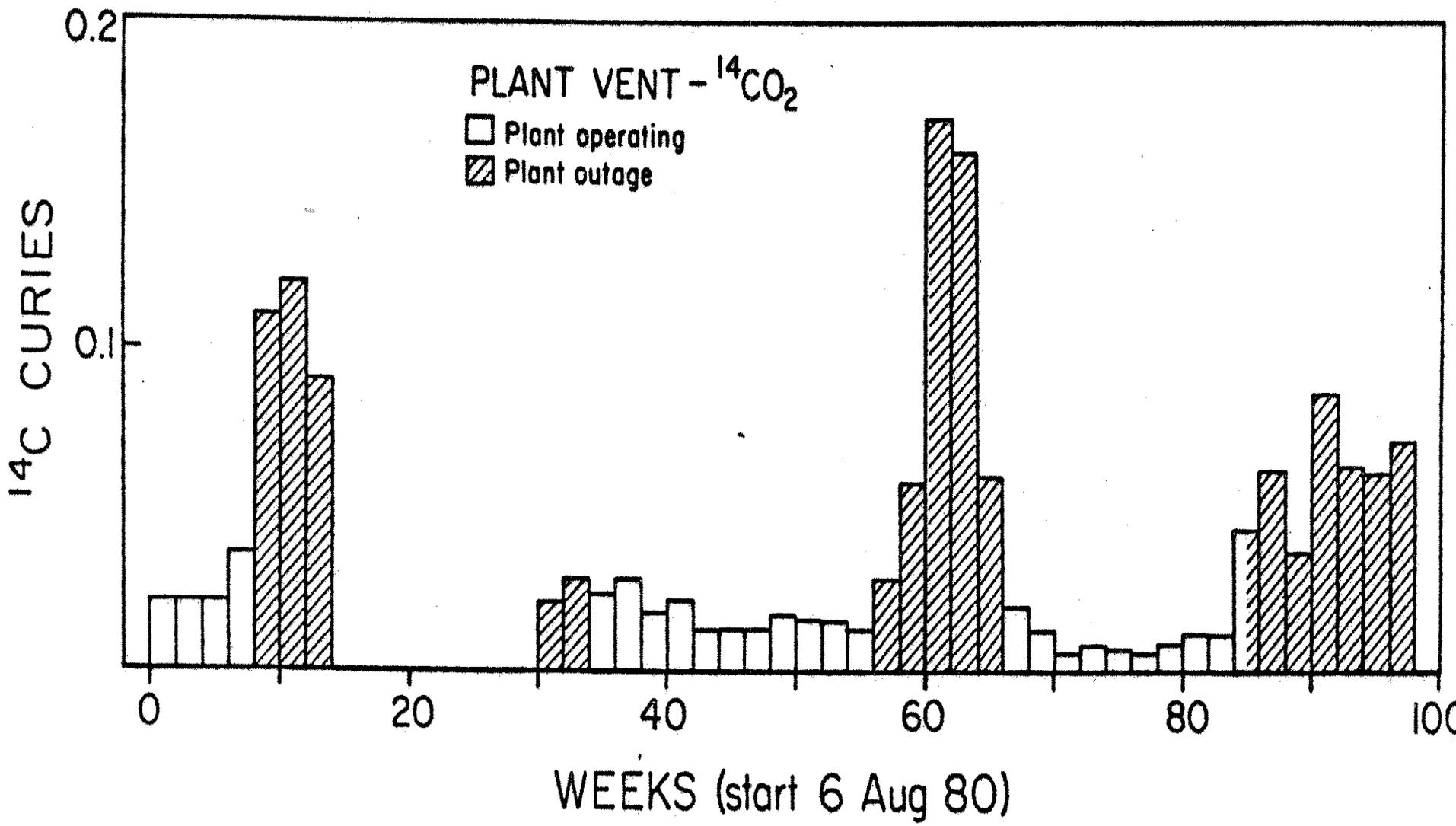
Other sample points

- Grab samples were taken from
 - Plant Vent
 - Containment Vent (during Vents / Purges)
 - Gas Decay Tanks (approx 7% of total)
- Cryogenic and gas chromatograph separation applied to isolate gases and count separately.
- Reactor Coolant samples were taken before and after the demin bed to try to determine a decontamination factor. These samples required complex radiochem preparation due to the inclusion of so many potential contaminants.

Integrating the C-14 releases (over 98 weeks)



Refueling Outages were by far the most significant interval for C-14 emissions.



Total Curies related to MW (e)

- Curie totals were compared to average MW produced, normalized annually.
- This was then corrected for rated MW (e).
- Release rate for total gaseous C-14 was
9.6 Ci/ GW (e) yr.

Operating vs Shutdown

- During operation, Hydrogen is the cover gas on the primary volume control tank.
- Under these conditions, carbon is reduced to organic forms (hydrocarbons, methyl and ethyl compounds).
- At shutdown and peroxide addition, carbon is oxidized to form CO₂.
 - This is also when the greatest volume of airborne activity is released.

Chemical Composition by Origin

GAS DECAY TANKS ($\mu\text{Ci/cc}$)

Collection Date	6/24/76	4/11/78	4/29/81	6/23/82
Chemical Species				
^{85}Kr	2.08 E-06 \pm 7%	2.76 E-04 \pm 5%	3.51 E-04 \pm 5%	2.34 E-04 \pm 5%
$^{14}\text{CH}_4$	2.0 E-05 \pm 10%	3.70 E-05 \pm 5%	2.81 E-05 \pm 5%	5.6 E-05 \pm 5%
$^{14}\text{C}_2\text{H}_6$	} 8.3 E-06 \pm 10%	} 2.62 E-05 \pm 5%	3.3 E-06 \pm 9%	1.10 E-05 \pm 6%
$^{14}\text{C}_3\text{H}_8$			6.9 E-06 \pm 6%	7.8 E-06 \pm 7%
$^{14}\text{C}_4\text{H}_{10}$			N.A.	4.1 E-06 \pm 7%
$^{14}\text{CO}_2$	1.15 E-06 \pm 5%	1.81 E-06 \pm 7%	1.32 E-05 \pm 5%	2.74 E-06 \pm 6%
Total ^{14}C	2.9 E-05 \pm 10%	6.5 E-05 \pm 5%	5.2 E-05 \pm 5%	8.2 E-05 \pm 5%

Avg: 62% CH_4 , 29% C_2H_6 , C_3H_8 , etc, and 9% CO_2
 C-14 from Gas Decay Tanks contribute approx 7% of total release, annually.

Chemical Composition by Origin

CONTAINMENT AIR
($\mu\text{Ci/cc}$)

(while operating)

<u>Collection Date</u>	<u>4/11/78</u>	<u>4/29/81</u>	<u>3/5/82</u>
<u>Chemical Species</u>	C-14 from vents and purges were ~40% of total (Mostly from Purges).		
^{85}Kr	1.85 E-06 \pm 5%	4.8 E-06 \pm 5%	1.23 E-05 \pm 5%
$^{14}\text{CO}_2$	1.93 E-07 \pm 5%	5.8 E-07 \pm 5%	3.4 E-07 \pm 5%
$^{14}\text{CH}_4$	1.79 E-06 \pm 5%	2.2 E-06 \pm 5%	6.1 E-06 \pm 5%
$^{14}\text{C}_2\text{H}_6$	}	6.0 E-07 \pm 5%	3.2 E-06 \pm 5%
$^{14}\text{C}_3\text{H}_8$		3.4 E-07 \pm 5%	4.7 E-07 \pm 5%
$^{14}\text{C}_4\text{H}_{10}$		7.7 E-08 \pm 5%	6.5 E-08 \pm 5%
<u>Total ^{14}C</u>	<u>2.9 E-06 \pm 5%</u>	<u>3.8 E-05 \pm 5%</u>	<u>1.02 E-05 \pm 5%</u>

Avg: 60% CH_4 , 32% C_2H_6 , C_3H_8 , etc, and 8% CO_2

This shows that C-14 was formed in the Rx Vessel and leaked into Containment. If the C-14 were formed in the AIR by neutron leakage, the predominant form would be CO_2

Chemical Composition by Origin

VENT AIR
($\mu\text{Ci/cc}$) No GDTs, no containment releases

<u>Collection Date</u>	<u>4/11/78</u>	<u>4/29/81</u>	<u>3/5/82</u>
<u>Chemical Species</u>			
^{85}Kr	<4 E-08	1.04 E-09 \pm 9%	8.2 E-10 \pm 12%
$^{14}\text{CO}_2$	1.2 E-09 \pm 44%	2.0 E-10 \pm 19%	1.8 E-10 \pm 21%
$^{14}\text{CH}_4$	1.03 E-09 \pm 14%	3.5 E-10 \pm 13%	3.4 E-10 \pm 25%
$^{14}\text{C}_2\text{H}_6$	} 3 E-10 \pm 80%	6 E-11 \pm 54%	8 E-11 \pm 36%
$^{14}\text{C}_3\text{H}_8$		8 E-11 \pm 25%	8 E-11 \pm 32%
$^{14}\text{C}_4\text{H}_{10}$		N.A.	6 E-11 \pm 28%
Total ^{14}C	2.5 E-09 \pm 36%	6.9 E-10 \pm 20%	7.4 E-10 \pm 26%

Avg: 46% CH_4 , 20% C_2H_6 , C_3H_8 , etc, and 34% CO_2

Organic C-14 from primary systems leaking into the aux bldg (VCT, etc), oxidized & released, making up ~ 15% of total.

Primary Coolant Removal Rate

- Samples taken before and after demin bed

PRIMARY COOLANT
($\mu\text{Ci/ml}$)

Collection Date	4/11/78		4/29/81		3/5/82	
	Inlet Cleanup Demin.	Outlet Cleanup Demin.	Inlet Cleanup Demin.	Outlet Cleanup Demin.	Inlet Cleanup Demin.	Outlet Cleanup Demin.
Inorganic ^{14}C	2.85 E-05 \pm 5%	3.34 E-05 \pm 5%	3.73 E-05 \pm 5%	3.22 E-05 \pm 5%	6.0 E-06 \pm 5%	4.8 E-06 \pm 5%
Organic ^{14}C	4.6 E-05 \pm 5%	3.48 E-05 \pm 5%	3.01 E-05 \pm 5%	2.80 E-05 \pm 5%	5.0 E-05 \pm 5%	5.8 E-05 \pm 5%
Total	7.5 E-05 \pm 5%	6.8 E-05 \pm 5%	6.7 E-05 \pm 5%	6.0 E-05 \pm 5%	5.6 E-05 \pm 5%	6.3 E-05 \pm 5%

- Essentially no removal of C-14 on the resin.

Primary Conclusion

New York Power Authority officially recognized that C-14 was a relatively important contributor to offsite dose.

Determined that it should be reported.

Secondary Conclusions

- The Plant Vent includes essentially all releases.
- Measured release of C-14 was 3.8 Ci/yr, with a significant amount of time with the Rx S/D.
- This corresponds to 9.6 Ci / GW(e) yr @ IP3.
 - Avg production rate in the coolant of previously tested PWRs was 7 Ci/GW(e) yr
 - Avg release rate of previously tested PWRs in Germany was 6 Ci/GW(e) yr.
- Releases from IP3 were 74% organic, 26% CO₂

Predominant Pathways

- Approximately 78% of the total releases were from Pressure Reliefs (vents) or Purges from Containment (mostly Purges at shutdown).
- Approximately 15 % was released from ventilating the aux building.
- Another 7% came from gas decay tanks.

Vents and Purges from Containment

- The bulk of all C-14 is released at the initial purge of containment following shutdown.
- After startup, C-14 releases are low until enough inventory builds up in the vessel and eventually, containment.

Liquid Effluents

- Discharge of C-14 in solid and liquid effluents was small compared to gaseous.
- No measurable decontamination factor on the resin bed.
- Using the concentration of C-14 in the primary coolant and an estimated 0.5 gpm makeup rate, the discharge rate would only be 0.07 Ci/yr.

Dose Calcs

- Not re-inventing new methodology.
- Using RG 1.109 with a few site specific pieces of data.
- Airborne Pathways included:
 - Vegetative Pathway
 - Child Ingestion
 - Child Inhalation
- Liquid pathway calculated despite very low values.

Airborne Pathway

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- Assumptions
 - 9.6 Ci release per GW(e)-yr, from DOH study, 1983.
After 1989, this value was updated to 2.5 Ci /yr,
representing only the portion that was CO₂ (26%)
 - $p =$ ratio of the time of C-14 releases to total time.
This value should represent an understanding of
photosynthetic activity only in daytime, or approx 0.5.
However, it was initially set to = 1.
 - $X/Q @$ Primary Receptor = $8.96E-7$ sec/m³
Determined from averaging 10 years of MET data
 - Child bone dose at the Primary Receptor is most limiting
 - Dose Factors used from Reg Guide 1.109

Airborne Pathway

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Vegetative Pathway (RG 1.109, App C)

$$C14 \text{ pCi/kg} = 3.17E7 \text{ (pCi/Ci)(g/kg)(yr/sec)} * p * Q * X/Q * (0.11/0.16)$$

$$Q = 2.50 \text{ Ci (initially, we used 9.6 !!!)}$$

$$P = 1 \text{ (Total release time per total time where photosynthesis can occur. No attempt was made to correct for photosynthesis only in daytime).}$$

$$X/Q = 8.96E-7 \text{ sec/m}^3 \text{ (using annual average MET data at garden)}$$

0.11=Default fraction of total plant mass that is natural carbon (unitless)

0.16=Conc of natural carbon in the atmosphere, g/m³
(not yet using site-specific values)

$$C14 = 49.3 \text{ pCi/kg ingested (veg)}$$

Airborne Pathway

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Child Ingestion

$$\text{Dose} = \text{DF} * [\text{uv} * \text{fg} + \text{uL}] \text{C14}$$

$$\text{DF} = 1.21\text{E-}5 \text{ mrem/pCi Bone Child}$$

$$2.42\text{E-}6 \text{ mrem/pCi TB Child}$$

$$\text{uv} = 520 \text{ kg/yr (child)}$$

$$\text{uL} = 26 \text{ kg/yr}$$

$$\text{fg} = .76$$

$$\text{Child Bone Dose} = 1.21\text{E-}5 * 20765 = 2.51\text{E-}01 \text{ mrem/yr}$$

$$\text{Child TB Dose} = 2.42\text{E-}6 * 20765 = 5.03\text{E-}02 \text{ mrem/yr}$$

Airborne Pathway

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Child Inhalation

$$C_{14} = 3.17E+4 * (Q) * (X/Q)$$

$$3.17E+4 = (\text{pCi-yr}) / (\text{Ci-sec})$$

$$C_{14} = 3.17E+4 \text{ pCi-yr/Ci-sec} * 2.5 \text{ Ci} * 8.96E-7 \text{ sec/m}^3 = 0.071 \text{ pCi/m}^3$$

$$\text{Dose} = \text{BR} * \text{Conc} * \text{DFi}$$

$$\text{BR}(\text{child}) = 3700 \text{ m}^3/\text{yr}$$

$$\text{DFI} = 9.70E-6 \text{ mrem/pCi Child Bone}$$

$$\text{Conc} = 0.071 \text{ pCi/m}^3$$

$$= 1.82E-6 \text{ mrem/pCi Child TB}$$

$$\text{Child Bone Dose} = 3700 * 0.071 * 9.7E-6 = 2.55E-3 \text{ mrem/yr}$$

$$\text{Child Tot Body Dose} = 3700 * 0.071 * 1.82E-6 = 4.78E-4 \text{ mrem/yr}$$

Total Summary

Total Annual Airborne Doses at Primary Receptor (nearest residence):

Child Bone = 0.254 mrem

Child Total Body = 0.0508 mrem

Liquid Doses

Assumptions:

- 0.07 Ci of C-14 released per GW (e)-yr, per DOH study, 1983
- C-14 is released to the river equally over the entire year
- Worst case curie source term assumed each year
- Avg Waste and Dilution volumes (1987 to 1999, as follows):
 - Average Annual Waste Volume = 1.33E6 gal
 - Average Annual Dilution Volume = 2.62E11 gal

$$\begin{array}{r} \text{Undiluted Concentration} \\ 0.07 \text{ Ci} \quad * \quad 1\text{E}6 \text{ uCi/Ci} \\ \hline 1.33\text{E}6 \text{ gal} \quad * \quad 3785 \text{ ml/gal} \end{array} = 1.39\text{E-}5 \text{ uCi/ml}$$

Liquid Effluent C-14 - worst case annual dose

WASTE VOLUME = 1.33000E+06 gal	DILUTION VOLUME = 2.62001E+11 gal
RELEASE DURATION = 525600 minutes	DILUTION FLOW RATE = 4.98480E+05 gpm
AVG RELEASE RATE = 2.530E+00 gpm	total MPCW eff = 3.000E-03
CURIES (C-14 only) = 7.000E-02 Ci's	BETA CURIES = 7.001E-02 Ci's

	PRE DILUTION uCi/ml	PRE DILUTION CONC/MPC	POST DILUTION uCi/ml	POST DILUTION CONC/MPC	MICRO- CURIES RELEASED
C-14	1.39E-05	4.64E-03	7.06E-11	2.35E-08	7.00E+04

-----< mrem >-----							
AGE	BONE	LIVER	TOTBOD	THYROID	KIDNEY	LUNG	GI-LLI
ADULT	4.16E-03	8.31E-04	8.31E-04	8.31E-04	8.31E-04	8.31E-04	8.31E-04
TEEN	4.53E-03	9.05E-04	9.05E-04	9.05E-04	9.05E-04	9.05E-04	9.05E-04
CHILD	5.83E-03	1.17E-03	1.17E-03	1.17E-03	1.17E-03	1.17E-03	1.17E-03

CRITICAL AGE GROUP : CHILD

CRITICAL ORGAN : BONE

Reporting the curies and dose

- Dose to Man Section of Reg Guide 1.21
- Included a narrative summarizing the study and our reported Ci and dose
- It has been updated a few times over the years, for example, when we realized we only want to report dose as a result of carbon dioxide.
- Last update was to apply lessons learned from Unit 3 to Unit 2, an identical unit, for IPEC.
- Included justification why it is reported separately and not with other dose/curie tables.

Total Dose:

In compliance with 40CFR190, the following table indicates the Total Dose, including any measured direct shine component from the site property for 2009:

		Whole Body	Max Organ
40 CFR 190 limit ==>	IPEC	25 mrem	75 mrem
Routine Airborne Effluents ¹	Units 1 and 2	2.28E-3	2.28E-3
Routine Liquid Effluents	Units 1 and 2	9.00E-4	1.71E-3
Routine Airborne Effluents ¹	Unit 3	3.36E-3	3.36E-3
Routine Liquid Effluents	Unit 3	2.49E-4	4.59E-4
Carbon-14 Totals (Liquid & Airborne releases from IPEC Units 1, 2, & 3)	IPEC	1.04E-1	5.20E-1
Ground Water & Storm Drain Totals	IPEC ²	2.56E-4	1.03E-3
Direct Shine from ISFSI, Radwaste Storage, SG Mausoleum, etc.	IPEC ³	5	5
Indian Point Energy Center Total Dose, per 40 CFR 190	IPEC	5.11	5.53

Investigating the high doses

- The RG 1.109-based values were quite high compared to published values of global C-14 burden.
- NCRP-76 (1984) : 6 pCi/g in human tissue.
- NCRP-81 (1985) : 1.25 mrem/yr, naturally, and approx 1% of this from Nuc Pwr Plants.
- New York Power Authority, Indian Point #3 decided to take a second look at the Kunz report and newer data, to determine if we should alter the way we are reporting C-14 releases and offsite dose.
- Contracted the Institute of Env Medicine at NY Univ Medical Center (Dr. Paul Linsalata), in 1989.

Findings from NYU, Institute of Environmental Medicine

P. Linsalata, Ph.D

- Curies were accurately measured and/or modeled from the Kunz study.
- We were very conservatively calculating the offsite dose component (and well within limits).
- Still, we could apply numerous site-specific data to be more accurate (lower) with our representation of offsite dose due to C-14.
- Some values would be difficult and expensive to generate and justify, but some - simple.

Suggested Improvements

from 1989

- Keep pathways (primarily ingestion and a slight amount from inhalation) but reduce the dose-driven source term to represent only the CO₂ portion (26% of total curies).
- Consider using more modern literature regarding the mean concentration of C-14 in the local atmosphere from a value of 0.16 g/m³ to 0.18 or 0.19 (Gammon, 1986).
- Fraction of total plant mass actually carbon is different for all vegetables/fruits. Site specific value may only involve corn. For the lower Hudson Valley, a more accurate value may be 0.075.

Suggested Improvements

2 of 4

- Local area has a maximum 6 month growing season, April to September (Killough and Rohwe, HP 34, 1978)
- Consider site specific ingestion rates, if they can be justified.
- Consider that photosynthesis occurs only in daylight.

Suggested Improvements

3 of 4

MET data:

- Annual Avg data appears to be very conservative when used for C-14.
- Look at joint frequency data (preferentially over many years) during DAYLIGHT and only during growing season.
 - IPEC calculated X/Q values during daylight and growing season are 50-80% of routine ann avg.

Suggested Improvements

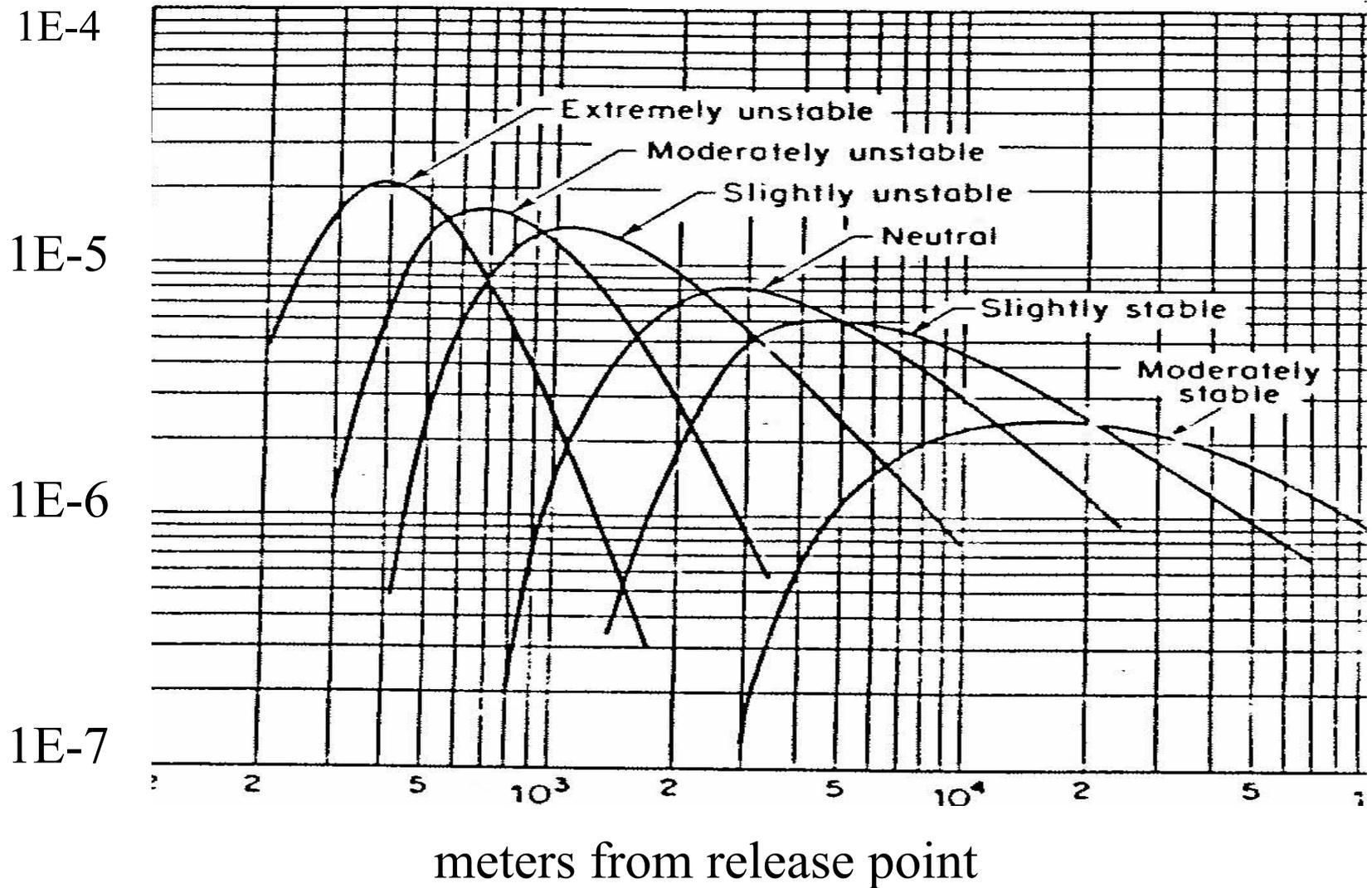
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MET data stability:

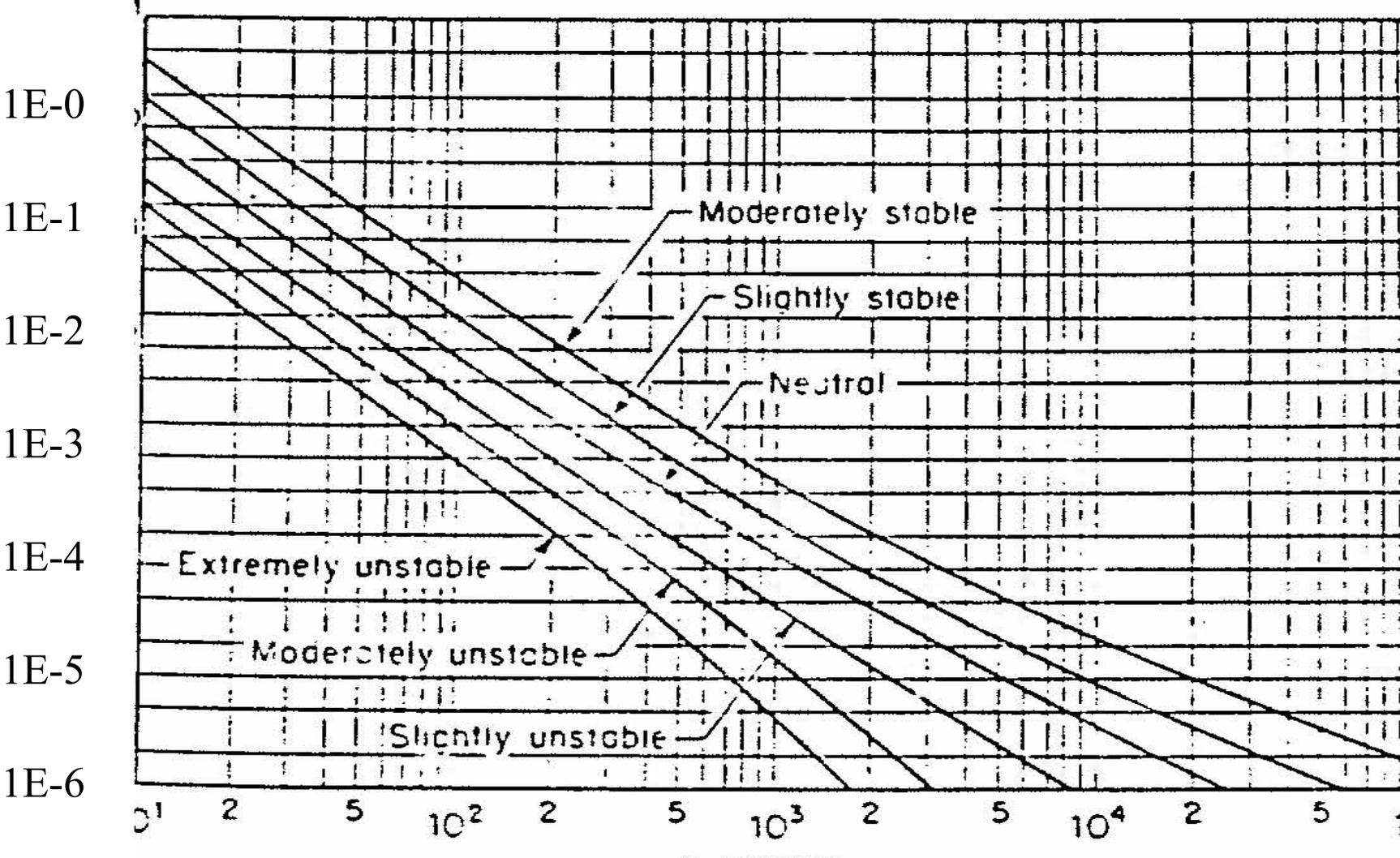
- Generally there is much more instability during the day. Since non-elevated X/Q goes DOWN with increased instability, lower X/Q is available.
 - Killough/Rohwer, HP, 1978 (used elevated only)
 - Eisenbud, 1987 and Hilsmeier & Gillford, 1962

X/Q values may increase with instability shifts at elevated release points, but at ground level, X/Q drops with increased instability.
- If doses are a real challenge, making the effort to collect and process joint frequencies during the appropriate interval may prove beneficial.

X/Qs from elevated release point, different stability



X/Qs from ground level release point, different stability



meters from release point

EPRI Radwaste Management Seminar, Boulder, Co Aug 1990

- New York Power Authority presented a summary from Dr. Linsalata's work.
 - “Evaluation of environmental parameters for estimating the local dosimetric consequences of routine gaseous C-14 emissions from PWR”
(Gary Re’).
- Only a few of the site specific suggestions were incorporated at the time for IP#3.

Few Improvements Incorporated

- Immediately applied only 26% of the C-14 curies released used for dose calculations.
- Collected more detailed MET data used for determining ODCM 10-yr average data, but we did NOT elect to apply any of the new special X/Q data.
- Limits were not challenged, and our chosen reporting method did not initiate further discussion, so the conservative assessment of 1990 (corrected for using only CO₂) was carried forward to 2004, eventually considered for IPEC's 2-unit site.

Updates to consider

In addition to updates to the model resulting in a curie value (per GW yr) from EPRI, IPEC is currently considering the following:

- Use of a 6 month growing season
- Recognition that there are not a lot of fruits/veg being grown or consumed locally (no grain), and that perhaps *corn* should be the sole source of the value for plant mass percentage that is carbon.
- Use modern literature defining our region to be closer to 0.19 g/m^3 for C-14 mean concentration in air (vice Reg Guide 1.109 default of 0.16).

Tweaks we probably will NOT consider, at least for now....

- Using ICRP 30 or later dose conversion factors.
 - We have used modern guidance before, but only when none was available in Reg Guide 1.109.
 - The improvements in calculated dose with newer literature are not significant enough to warrant the risk of negative consequences of “cherry picking”.
- Using a specially prepared X/Q generated in daylight hours only during the growing season.

Guidance for use of these and other dose calculation parameters is forthcoming from the NEI portion of the C-14 task force.....

Veg Dose equations

- Old version:

$$C14_{pCi/kg} = 3.17E7 * p * Q * X/Q * (0.11/0.16)$$

$$Q = 2.50 \text{ Ci}, \quad P = 1, \quad X/Q = 8.96E-7 \text{ sec/m}^3$$

- Potential new version:

$$C14_{pCi/kg} = 3.17E7 * p * Q * X/Q * (0.075/0.19)$$

$$Q = 2.50 \text{ Ci}, \quad P = .25, \quad X/Q = 8.96E-7 \text{ sec/m}^3$$

- P represents a growing season of half the year, 12 hrs/day
- 0.075 local fraction of total mass of veg that is carbon
- 0.19 g/m³ is local concentration of natural carbon in atmosphere

A reduction in dose by an approximate factor of seven

Indian Point from the west

Questions ?

